



Energy &
Homeland Security

Sustainable Aviation Fuel R&D at Sandia National Laboratories >>>>>

Presented by

Anthe George: angeorg@sandia.gov

Ryan W. Davis: rwdavis@sandia.gov

Alexander Landera: alande@sandia.gov

Isaac Ekoto: iekoto@sandia.gov

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SAND2023-12144C



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graph LR; A[Introduction] --> B[Fuel Production]; B --> C[Fuel Identification]; C --> D[End-Use]; D --> E[Emissions & Repercussions]
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Introduction

Fuel
Production

Fuel
Identification

End-Use

Emissions &
Repercussions



Energy &
Homeland Security

Fuel Production

Ryan W. Davis
Principal Member of Technical Staff



SANDIA'S LEGACY IN ADVANCED BIOFUELS PRODUCTION

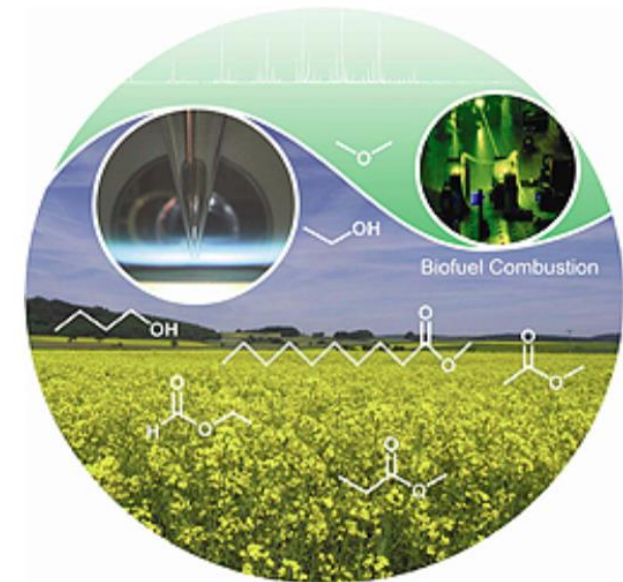
- Sandia staff have engaged in biofuels R&D for over 3 decades
 - in-house collaboration between Combustion Research Facility (CRF) and Chemical and Biological Sciences.



- Examples of recent production process development and scale-up for heavy-duty ground transport

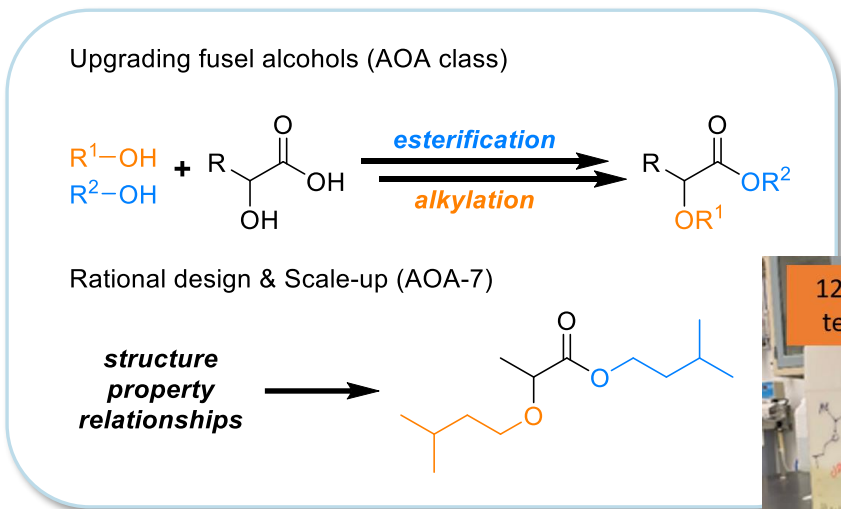
- Opportunities for utilizing new biomass feedstocks for SAF

- Lessons from sustainability and cost assessments

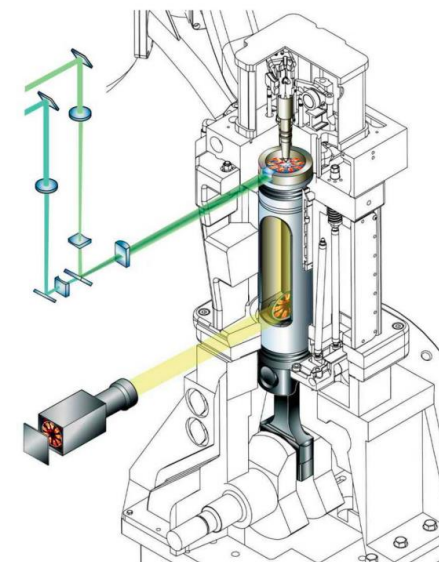
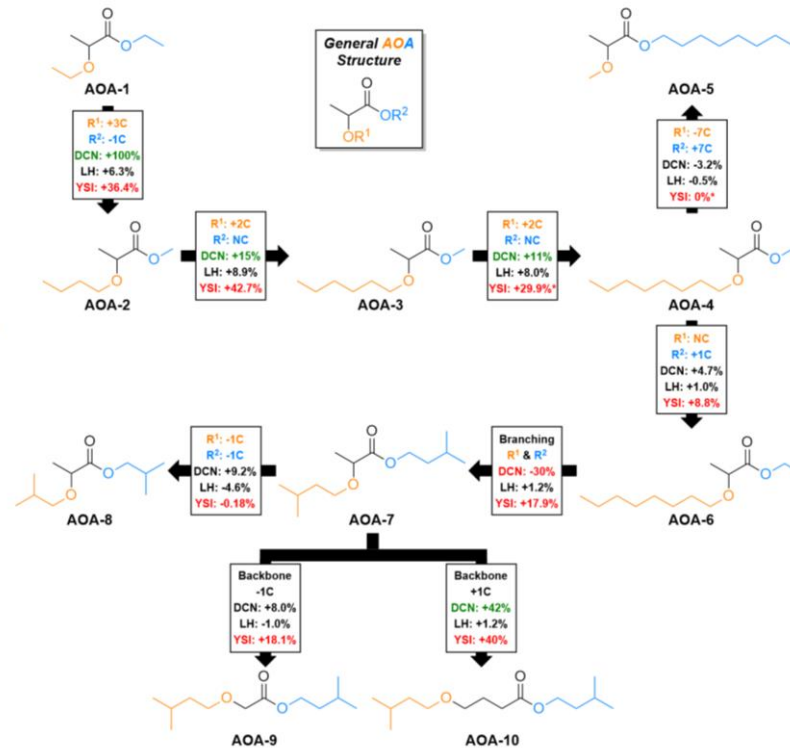
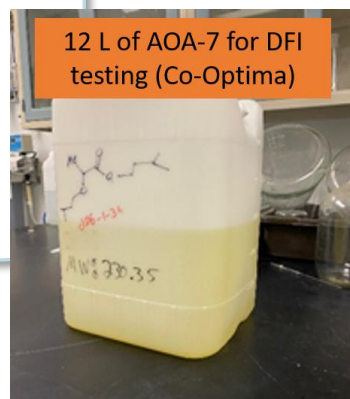




Demonstrated Capabilities – From feedstock to production for engine testing



Monroe et al *Energy & Fuels* 2022



Nilsen et al *SAE Intl* 2019

We are using computational modeling to identify promising SAF production targets for adherence to JetA specifications, and identifying opportunities for reduced sooting

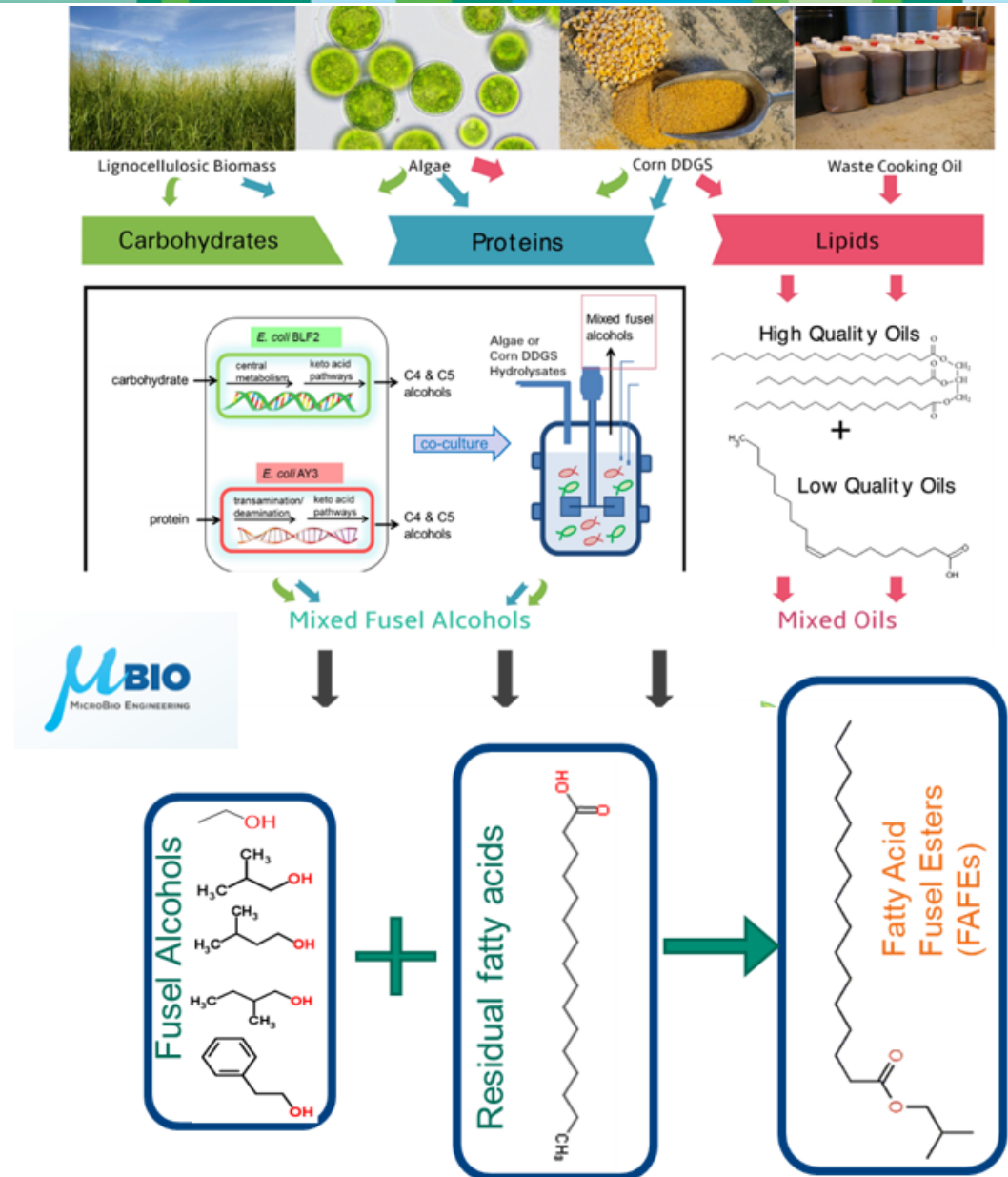


Co-Optimization of Fuels & Engines



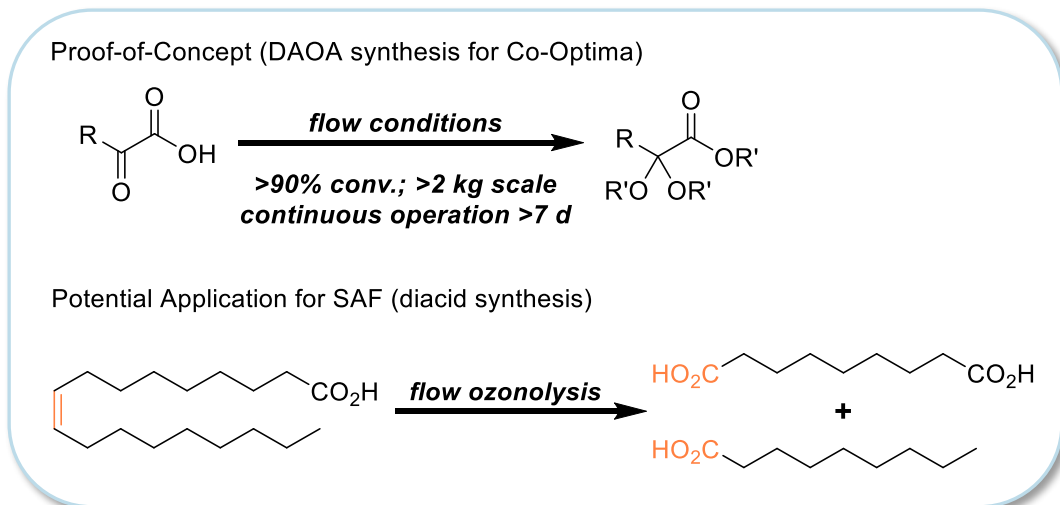
- Production process maximizes utilization of mixed substrates commonly found in algae, food waste, and distiller's grains, e.g. roughly equal fractions of lipids, carbs, and proteins
- FAFEs provide *improved LHV (+15%), cetane (+21%), and cold flow (-7°C cloud point) performance* compared to FAME, without sacrificing viscosity, lubricity, or sooting metrics
- Production methods compatible with existing biodiesel production and distribution infrastructure
- Subsequent reduction of ester to ether, i.e. fatty alkyl ethers, provides additional MCCI fuel performance, including further *improved cold flow (-15°C cloud point), LHV (+11%), and sooting (-7.4 YSI/MJ)*

Klein et al *Biomass Bioenergy* 2021
 Monroe et al *Fuel* 2020
 Carlson et al *Energy & Fuel* 2020



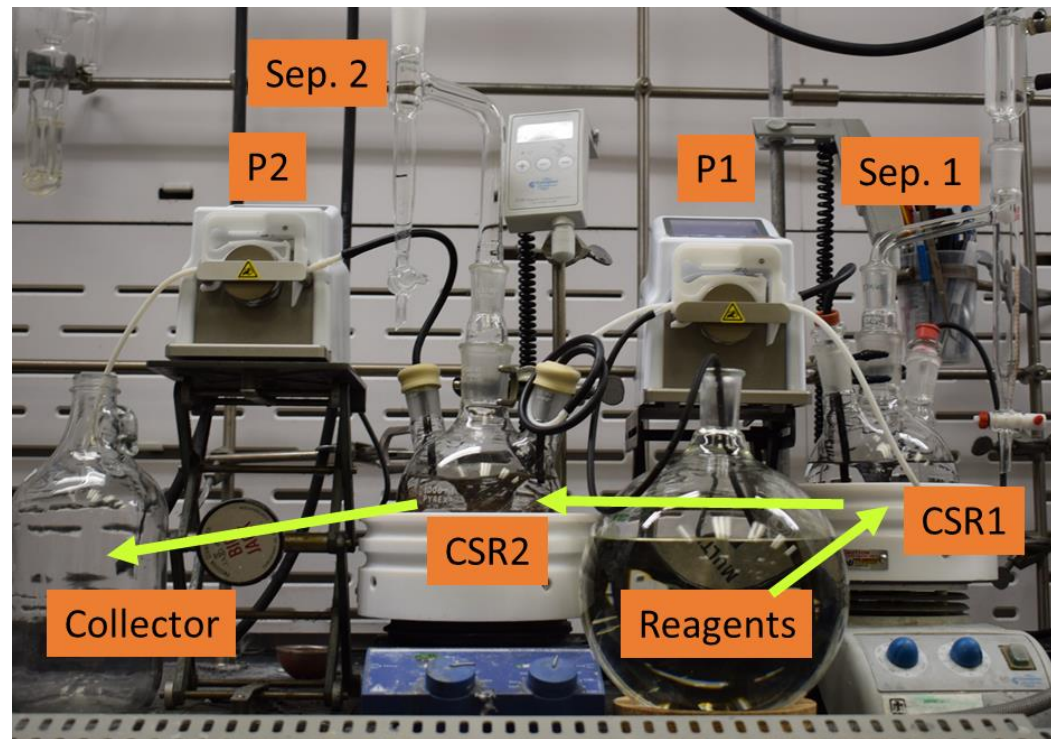


Demonstrated Capability – Continuous flow chemistry



Myllenbeck et al. manuscript in preparation for *Green Chemistry*.

Myllenbeck N, Davis RW, Monroe E, Carlson J "Alkyl-dialkoxoalkanoates as bioderived, high cetane diesel fuels" US Patent No. 11,492,565

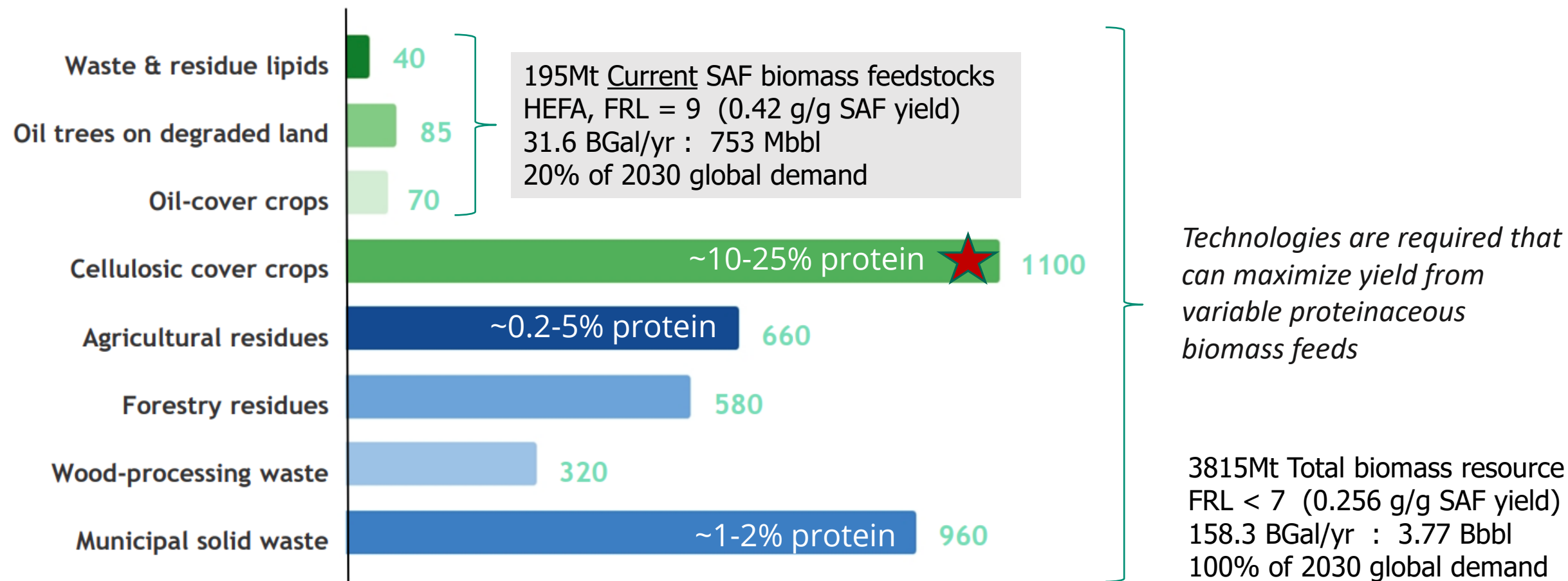


We have facilities and expertise for lab-scale conversion testing and process integration using next generation biomass feedstocks

OPPORTUNITY: UNDERUTILIZED PROTEINACEOUS BIOMASS FEEDSTOCKS CAN INCREASE PRODUCTION CAPACITY BY 400% TO FULFILL THE SAF GRAND CHALLENGE



2030 Practical Feedstock Availability (Mt)





Guiding Principles:

- Current feedstocks for high TRL/FRL SAF production (HEFA, AtJ, Biomass gasification) can provide up to ~20% of the 2030 SAF demand.
- Physico-chemical and combustion properties of final products must adhere to tiered-screening req's for SAF (JetA)
- Vetted pathways focus on those providing sufficiently low carbon intensity and cost-efficient production, with special attention to realizable yield, H₂ consumption, petroleum refinery integration, and feedback from SAF industry

Technology Concept: Mixed Proteinaceous Biomass to Fusel Alcohol SAF intermediates

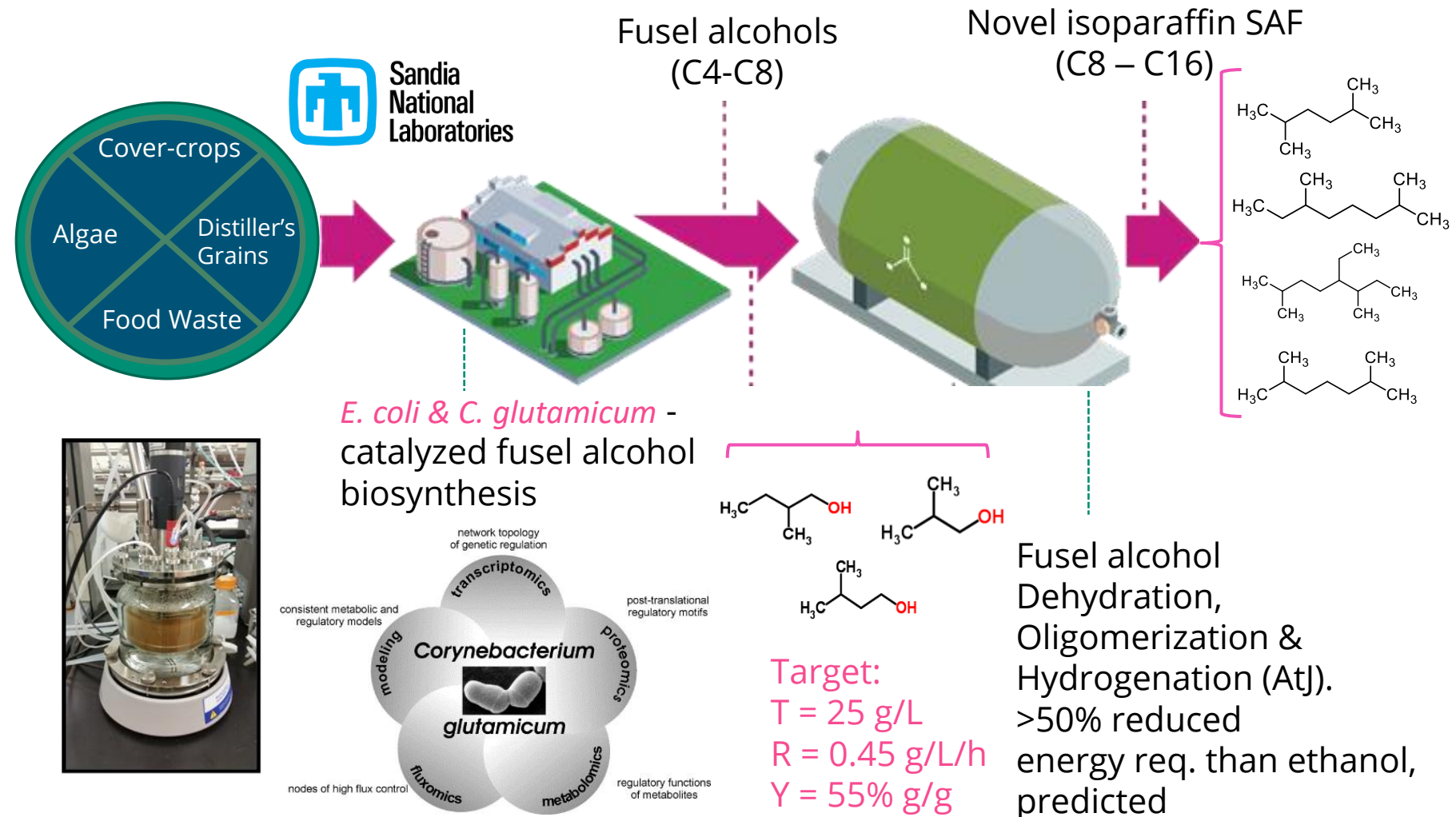
- *Current regulatory limits for fusel alcohol co-products of bioethanol would support ~4% of US SAF demand (18M bbl). By providing the capability to obtain >26% w/w conversion yield from cover crops, we can provide up to 38% of the total US SAF demand.*



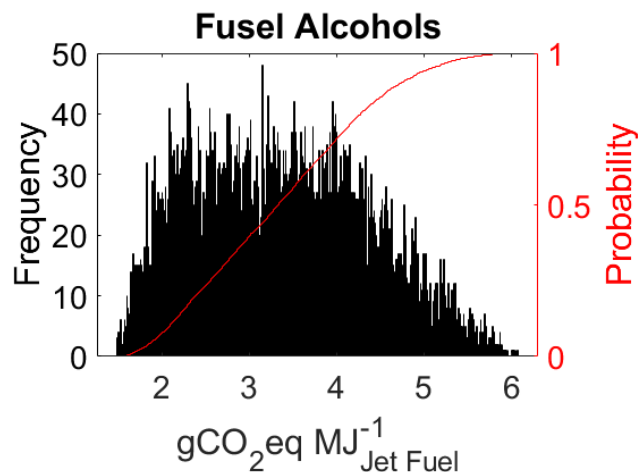
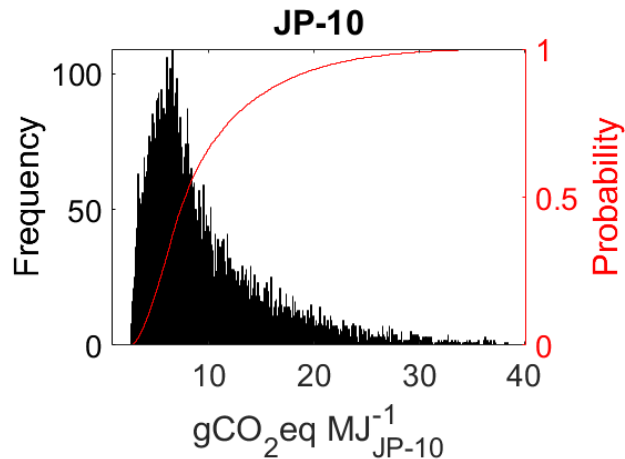
Novel Fusel-SAF Blend for Net Zero Emissions

Goal: to de-risk bioconversion of proteinaceous biomass for fuels and co-products for GHG savings

Sandia seeks partners to intensify processes for new biomass feedstocks



LIFE CYCLE CO₂ EMISSIONS ESTIMATES SUGGEST THAT FEW SCENARIOS PROVIDE TARGET GHG REDUCTION COMPARED TO CRUDE OIL-DERIVED JETA



- LCA assuming energy and materials variation by **+/- 20%**, and **H₂ consumption uncertainties**,
- **JP-10** can achieve life cycle **CO₂eq emissions at or below 3.6 gCO₂eq MJ⁻¹_{JP-10}** through process optimization, e.g., Lower H₂ consumption at 0.14 KgH₂ Gal_{Jet fuel}
- **Fusel alcohol AtJ** can achieve life cycle **CO₂eq emissions at or below 1.9 gCO₂eq MJ⁻¹_{Fusel alcohol}** through process optimization, e.g., Lower H₂ consumption at 0.08 KgH₂ Gal_{Jet fuel}

Best cases for minimizing CO₂ emission correspond to reduced H₂ requirement

Other authors

Conventional Jet fuel from crude oil: **11.1 gCO₂ MJ⁻¹**

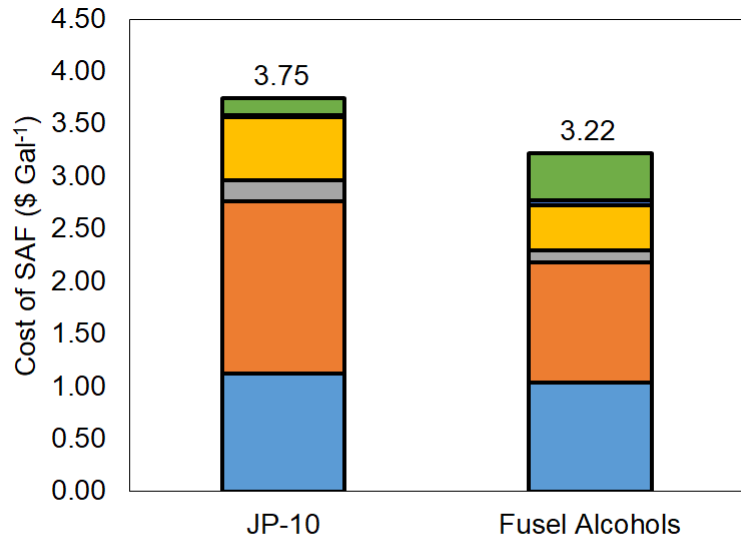
Corn oil-based renewable Jet fuel: **22.6 gCO₂ MJ⁻¹**

Fischer-Tropsch Jet fuel from Biomass: **4.5 gCO₂ MJ⁻¹**

PRELIMINARY TEA OF SANDIA'S SAF CONCEPTS INDICATES COMPETITIVE COSTS COMPARED TO EXISTING METHODS



Fuel cost at *current* 2.8\$/kg H₂:



- | | |
|------------------------|-------------------------|
| ■ Project capital cost | ■ Raw material cost |
| ■ Utilities cost | ■ Operating labor cost |
| ■ Maintenance cost | ■ Depreciation expenses |

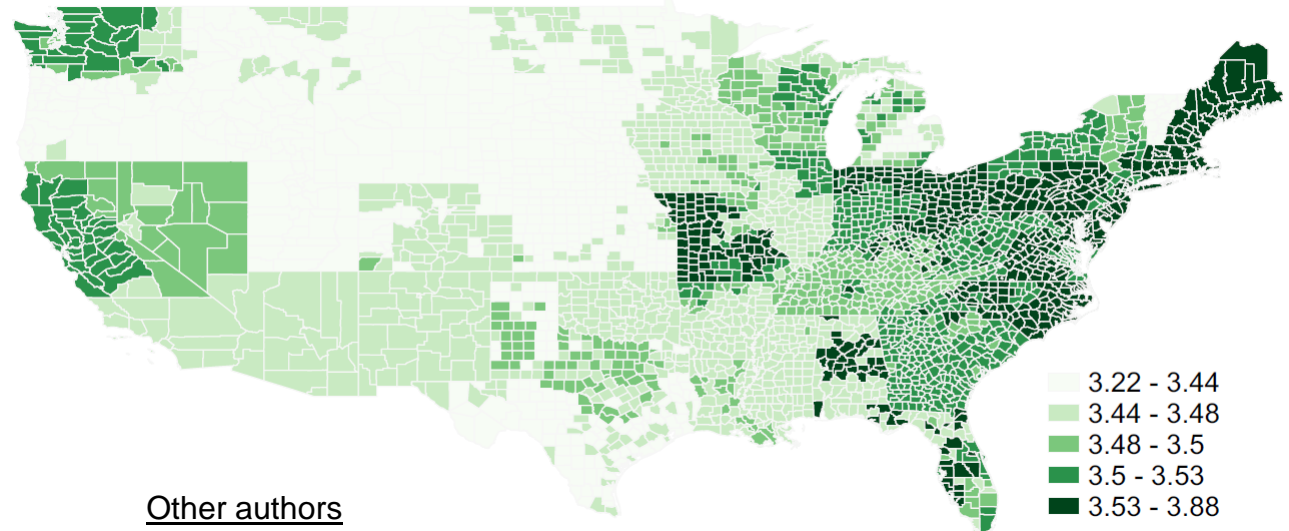
Assumptions

Electricity cost: 0.07\$/KWh

Feedstock costs:

- 101 \$/ton waste
- 85 \$/ton lignocellulosic biomass
- 1.13 \$/Gal methanol
- 1225 \$/ton toluene
- **2.8-6.1 \$/kg hydrogen**

H₂ is a major driver of Jet fuel cost (e.g., Fusel alcohols in \$/Gal_{Jet Fuel}):



Other authors

Hydrotreated depolymerized cellulosic jet:

- Corn stover: **6.20 \$/Gal**
- Wheat straw: **7.82**
- Sugar cane: **5.71**
- Forestry residues: **5.22**

Hydrogenated esters and fatty acids:

- **3.70 to 11.83 \$/Gal**



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Fuel Identification

Alexander Landera
Member of Technical Staff





- SAF physical property values measure safety, and performance of a SAF
- A drop-in SAF is highly desirable
 - No changes to aircraft infrastructure needed
 - Cannot be drop-in if it does not meet ASTM standards
- Estimating physical properties of a SAF can
 - Help determine issues early
 - Identify promising SAF components
 - Establish blend limits
- Blend models are hard to develop
 - Large number of components
 - Sparse data parameters
 - Time to solution must be fast

Physical property	Constraint
Specific Energy, MJ/kg	> 42.8
Energy Density, MJ/L	***
Density at 15°C, kg/m³	775-840
Flashpoint, °C	> 38
Melting point, °C	< -40

Physical property metrics, their constraints,
and Jet-A median values
From ASTM D-7566



Approach: Employ modeling in place of time-consuming laboratory measurements for screening SAF components and blends.

Modeling methods employed:

- Equation of State
 - Group contribution theory
- Quantum chemistry
 - Structure energy
 - Reaction barriers
 - Optimized geometries
 - Enthalpies of reaction

Quantities predicted:

Solid-Liquid-Equilibrium

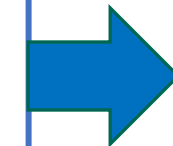
Energy density/Specific energy

Vapor pressure

Liquid densities

Flashpoint

Liquid viscosities



Tier α pred.
Important for
cold flow and safe handling

Reaction rates

Soot production

Polymer swelling (o-ring material)

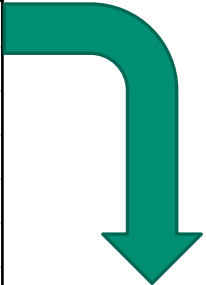


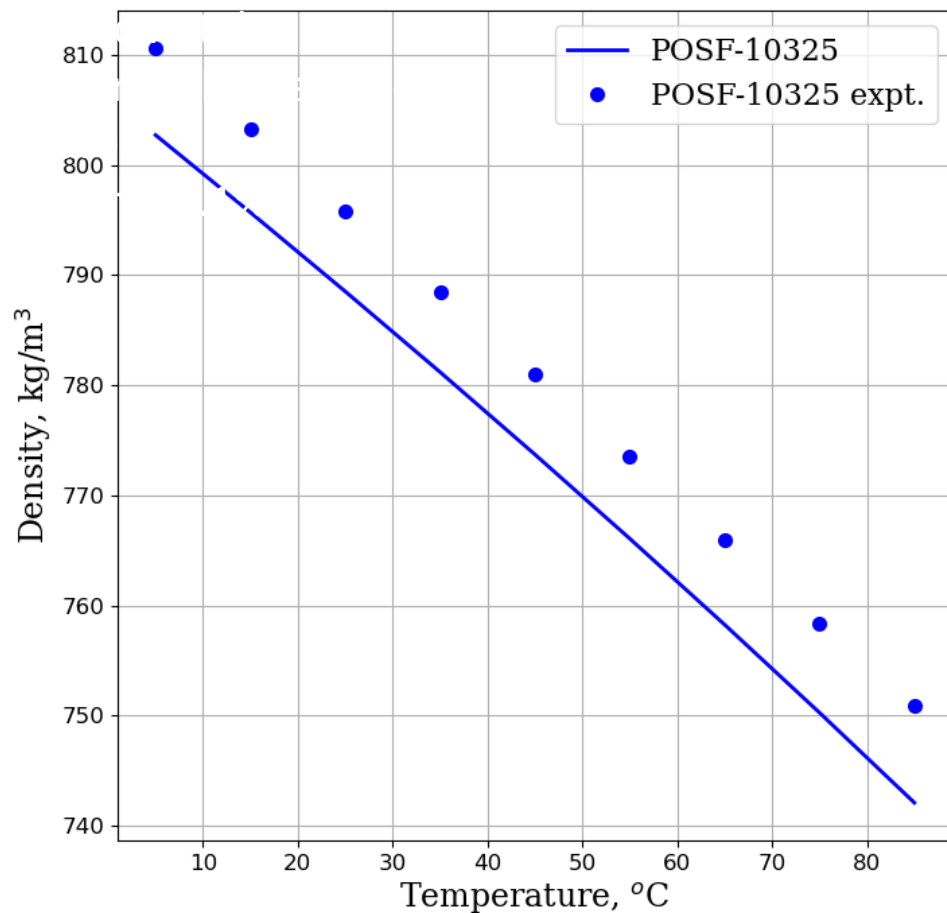
Goal: Use GCxGC data to accurately estimate physical properties of reference aviation fuels

Method: Eliminate minor chemical species (those which are present < 1 wt%)

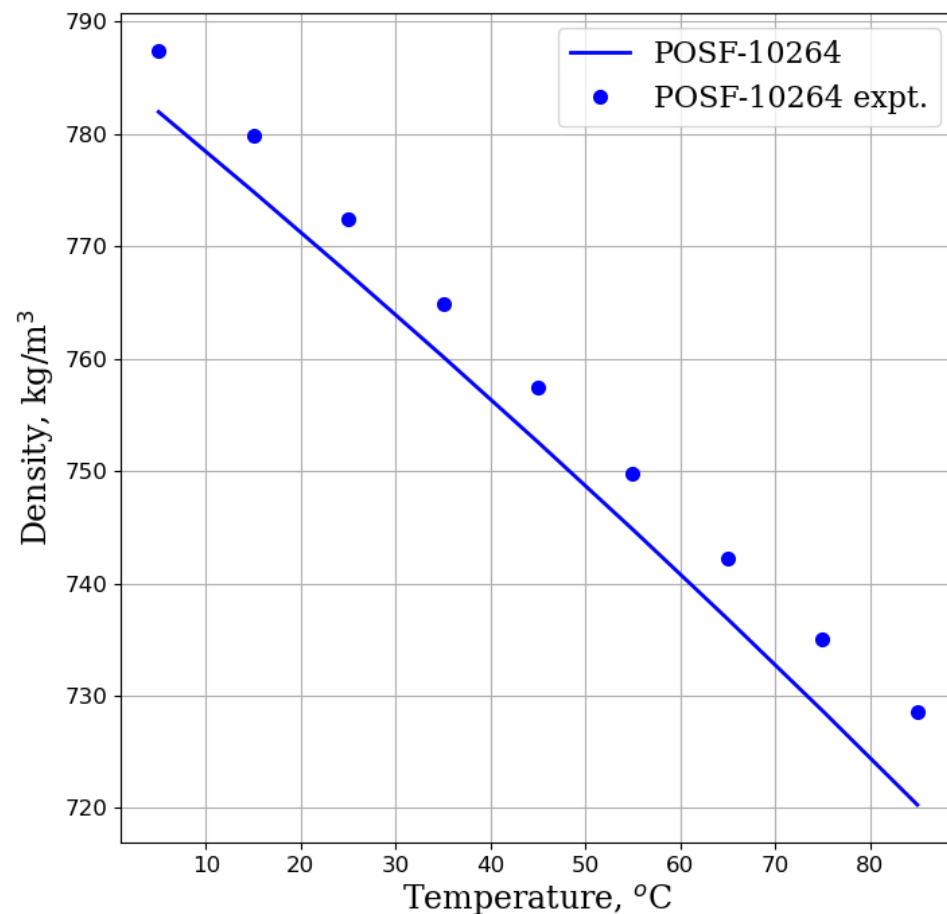
- Focus on chemical classes for which enough data is available
 - Branched alkanes => 2-methylalkanes
 - Alkyl benzene => linear alkyl benzenes
 - Alkyl monocycloalkanes => linear alkyl cycloalkanes
- These decisions are based on the availability of data, not on actual isomers present in the reference fuels

	Weight %	Volume %
Aromatics		
Alkylbenzenes		
benzene (C06)	0.01	0.01
toluene (C07)	0.23	0.21
C2-benzene (C08)	1.98	1.77
C3-benzene (C09)	4.17	3.73
C4-benzene (C10)	2.33	2.09
C5-benzene (C11)	1.19	1.07
C6-benzene (C12)	0.66	0.59
C7-benzene (C13)	0.25	0.22
C8-benzene (C14)	0.12	0.11
C9-benzene (C15)	0.06	0.05
C10+-benzene (C16+)	<0.01	<0.01
Total Alkylbenzenes	11.00	9.85

- 
- Ethylbenzene
 - Propylbenzene
 - Butylbenzene
 - Pentylbenzene



Density of POSF-10325



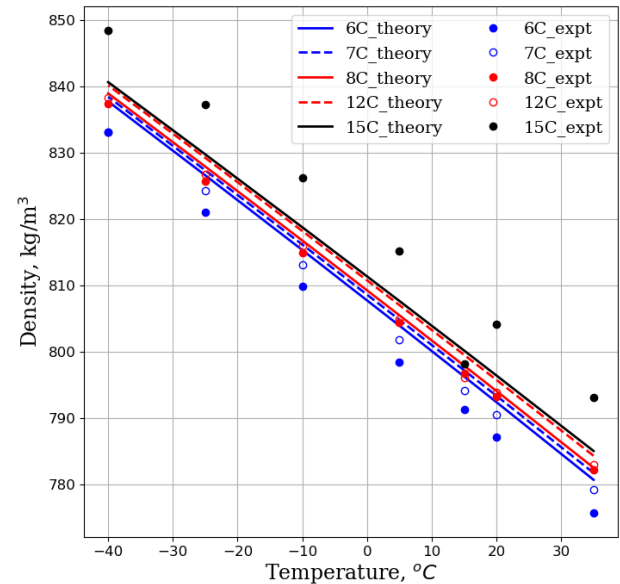
Density of POSF-10264

Accurate densities within ~ 1% are obtained

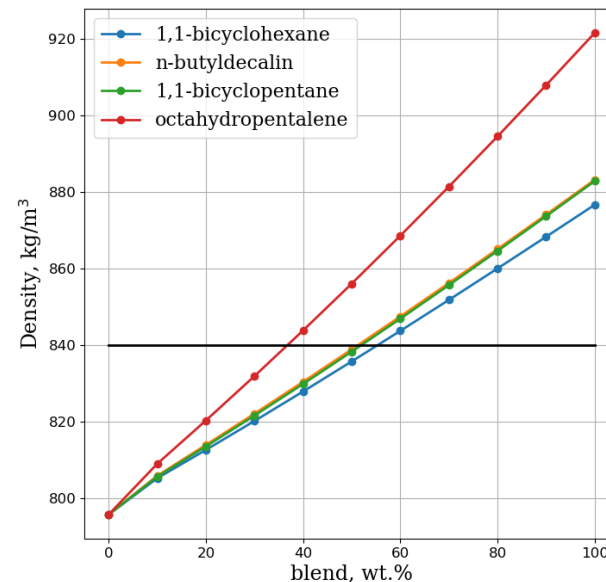
DENSITY PREDICTIONS ARE IN GOOD AGREEMENT WITH EXPERIMENTAL MEASUREMENTS



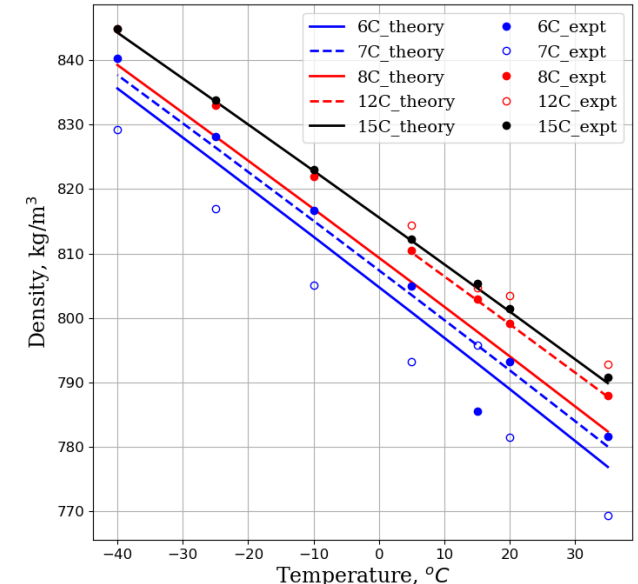
- Blend model shows that density can be accurately modeled. Largest error in validation is 1.4%
- Validated with **C6-C15** cycloalkanes
- A select group of cycloalkanes were chosen
 - Based on not meeting density requirements
- Blend models show that density can be accurately modeled
- Blend model shows that even though these molecules can't be used neat, they can be used as blends in about 35-60 wt.%
- Experimental measurements were performed in collaboration with Los Alamos National Laboratory



Density of 10 wt.% blends in jet-A (10325)

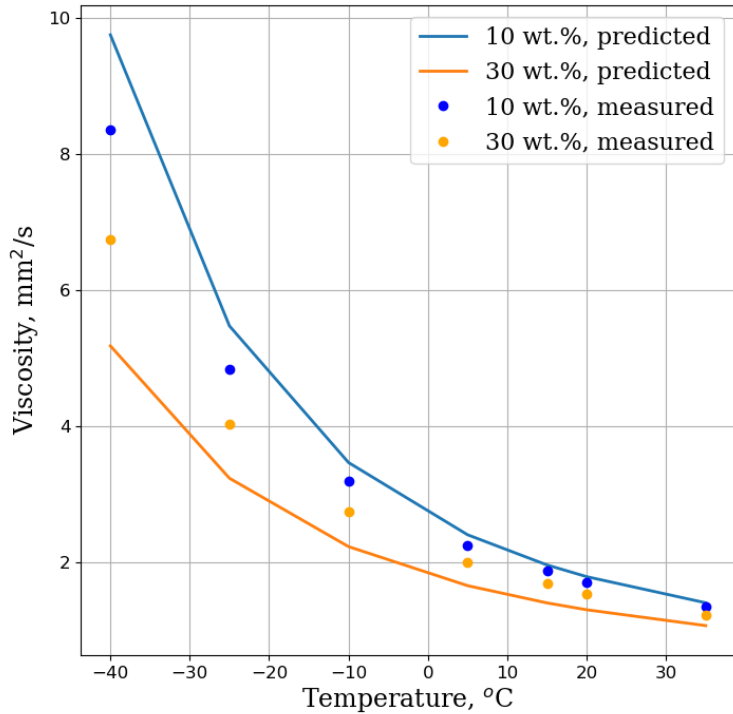


Density of blends in jet-A (10325)

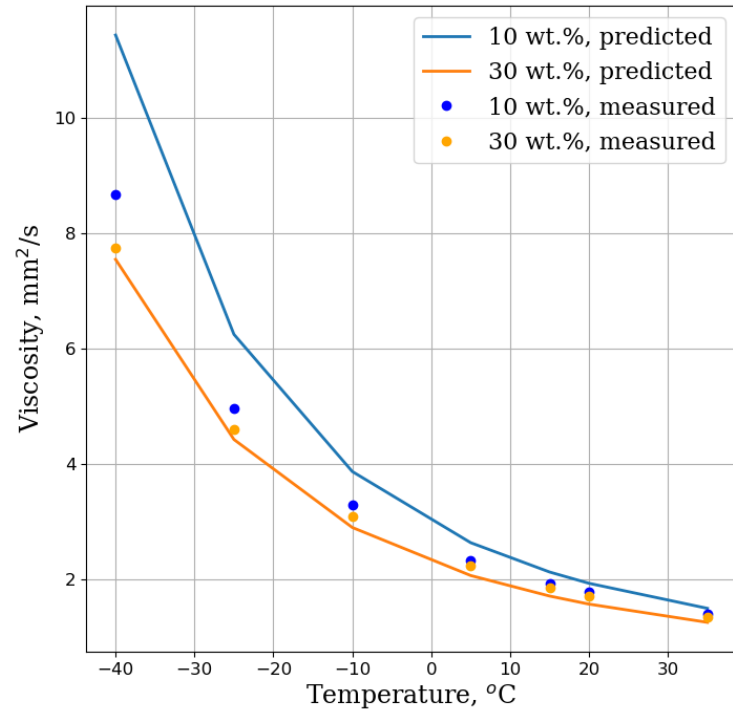


Density of 30 wt.% blends in jet-A (10325)

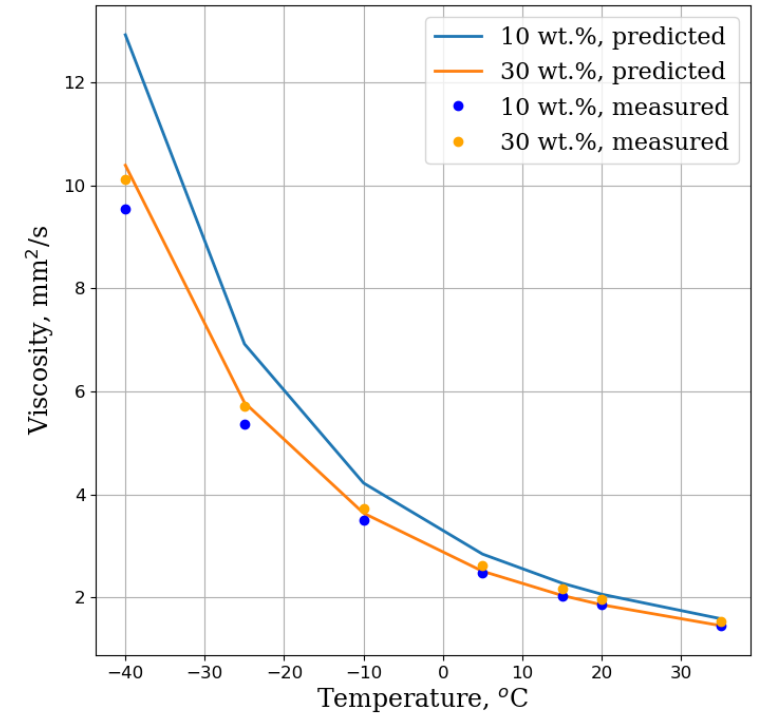
Molecules can be blended in 35-60 wt.% in Jet-A



POSF-10325 + cyclohexane



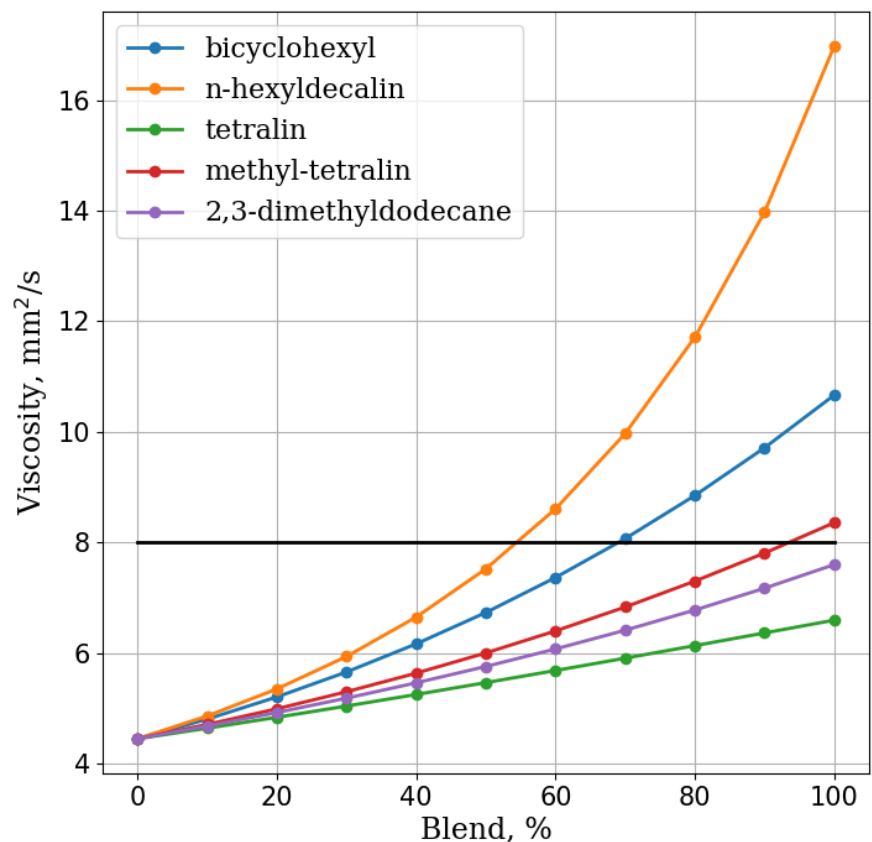
POSF-10325 + cycloheptane



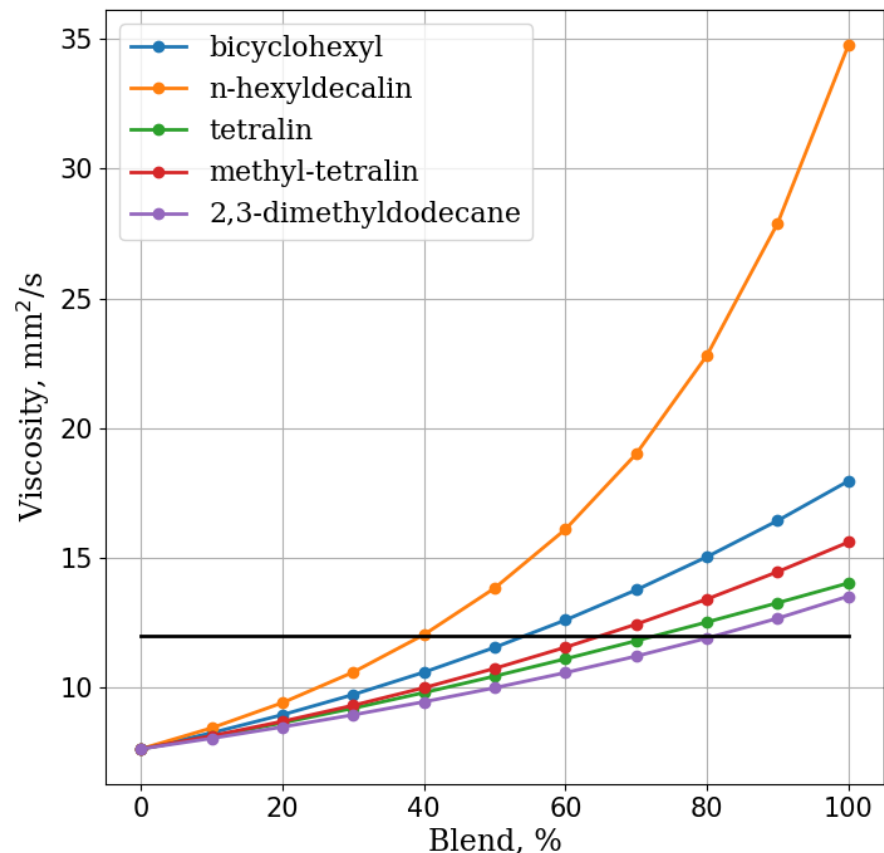
POSF-10325 + cyclooctane

Experimental viscosities are in good agreement with our modeled viscosities
This is a good initial validation step

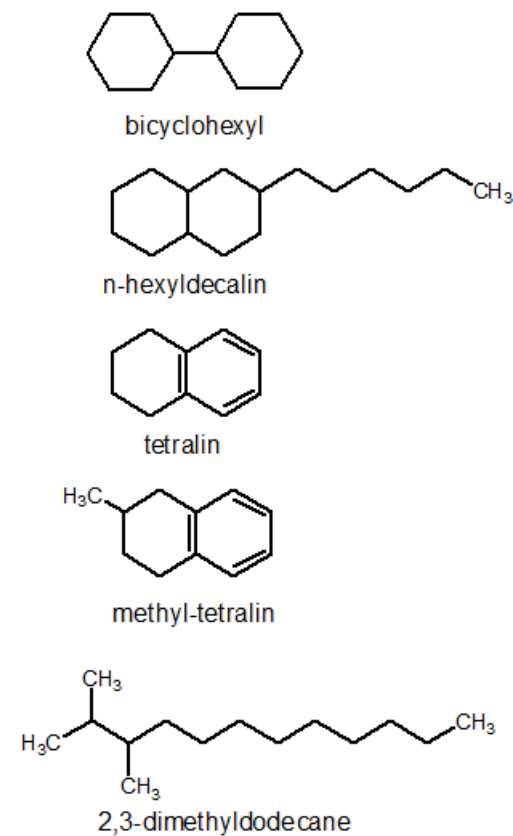
SELECTED CYCLOALKANES MEET VISCOSITY REQUIREMENT AS A BLEND OF UP TO 40-80% BY WEIGHT



Viscosity of cycloalkane + POSF-10325 at -20°C



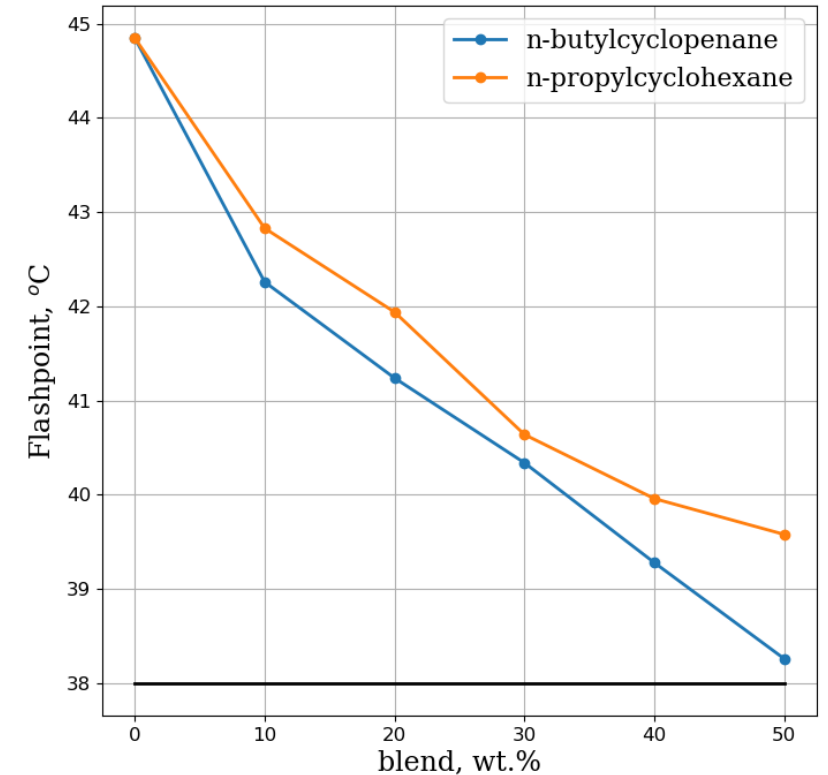
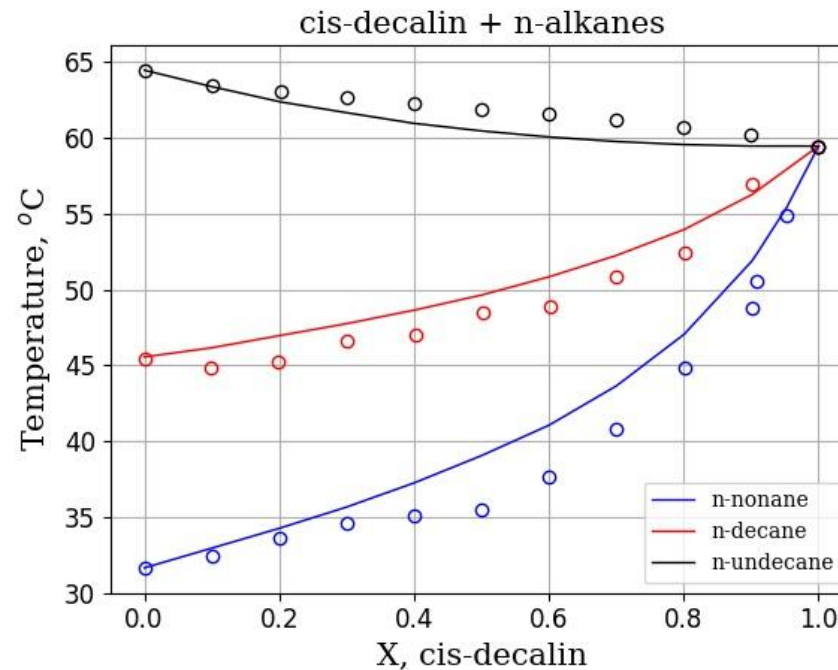
Viscosity of cycloalkane + POSF-10325 at -40°C



Meeting viscosity requirements at -20°C does not ensure you meet requirements at -40°C
Shows the blending limits of select, non-validated, alkanes in POSF-10325

FLASHPOINT IS ACCURATELY PREDICTED USING A SIMPLE BLEND MODEL

- Flashpoints are accurately predicted using a simple blend model
- N-butylcyclopentane and n-propylcyclohexane were chosen as representative cycloalkanes for blend study
- As neat molecules they do not meet the flashpoint requirements for jet-A fuel
- Flashpoint decreases steadily as blends increase
- n-butylcyclopentane and n-propylcyclohexane can be blended to at least 50% by wt.

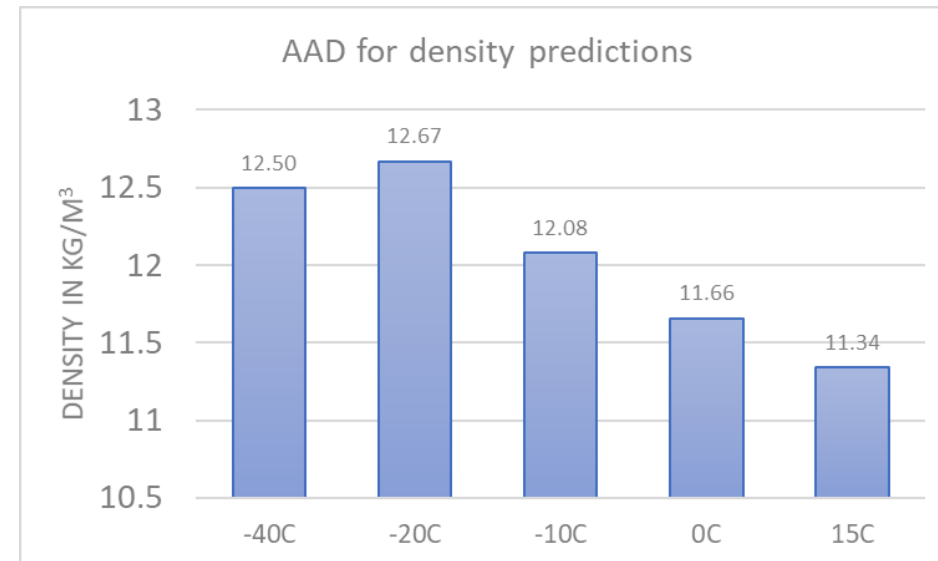
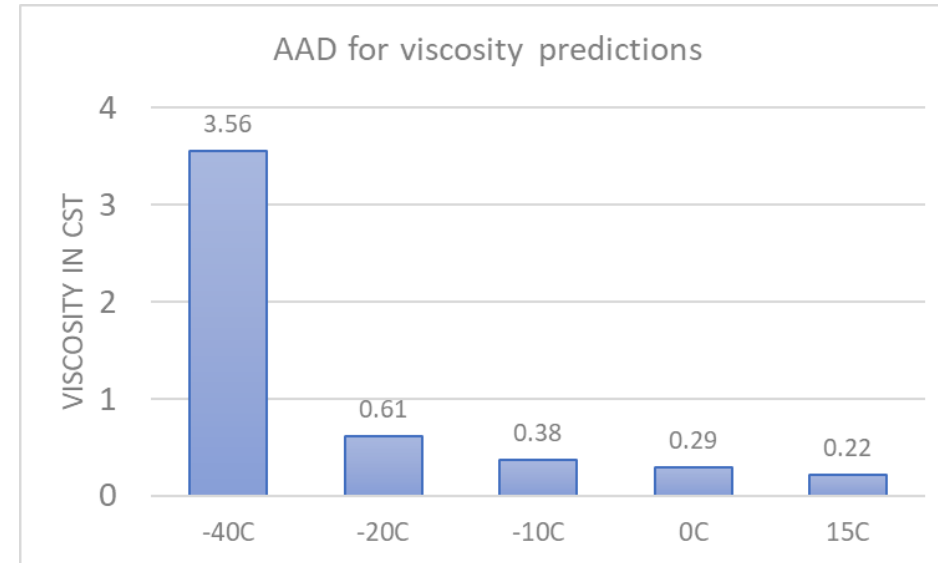


VISCOSITY OF COMPLEX FUELS (WITH U. DAYTON)

U. Dayton analyzed complex jet fuels

- Viscosity and Density were meas.
- Temp from -40C to 15C
- Total of 63 fuels were measured
 - From 2020 to 2021
- Less fuels at -40C due to freezing

- Viscosity predictions => SUPERTRAPP
- Density predictions => Ratchet eq.
- Viscosity at -40C is difficult to handle
- Other temperatures are much better



CAN CYCLOALKANES REPLACE AROMATICS?

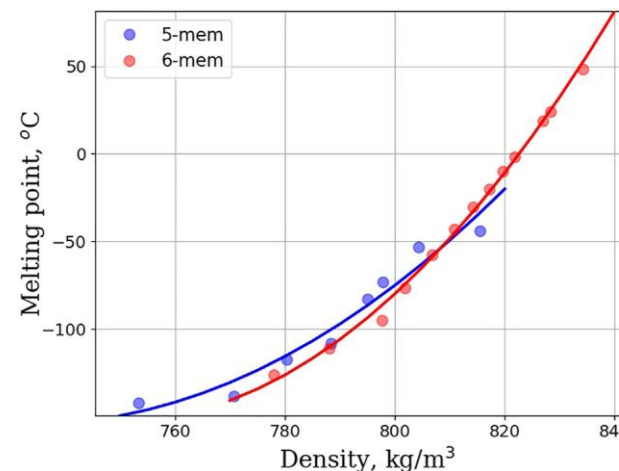
The challenge: aromatics promote seal swelling but have high sooting tendencies

Approach: conduct a modeling study and literature review to address the following question:
Are cycloalkanes viable replacements for aromatics?

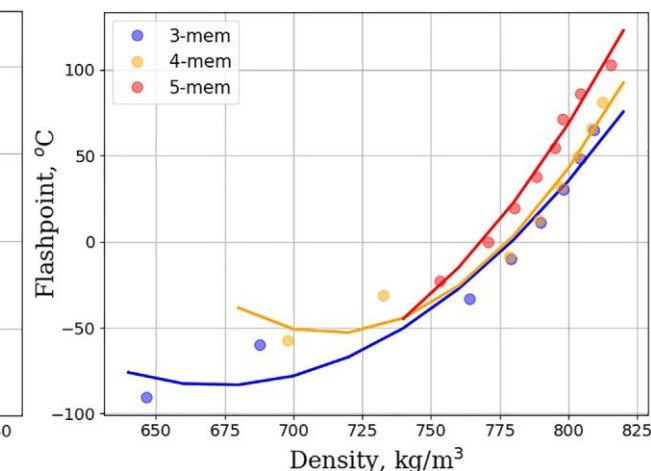
Our findings:

- Cycloalkanes with single substitutions generally have the best physical properties.
- Unsubstituted cycloalkanes suffer from high melting points
- Monosubstituted cycloalkanes have beneficially low melting points
- Poly-substitution typically leads to higher soot levels.
- Flat fused cycloalkanes (e.g., decalins) are good seal swelling agents

It depends!



Melting point of monosubstituted 5 and 6 membered rings with linear alkane substitutions.



Flashpoint: monosubstituted 3,4, and 5 membered rings with linear alkane substitutions.

Landra A., et al “Building Structure-Property Relationships of Cycloalkanes in Support of Their Use in Sustainable Aviation Fuels”. *Front. Energy Res.*, 2022, 9:771697. doi: 10.3389/fenrg.2021.771697



High performance fuel components:

- Strained ring structures for enhanced energy density and specific energy

Cycloalkane properties:

- Selecting suitable replacements for conventional aromatics

Development of new blend components:

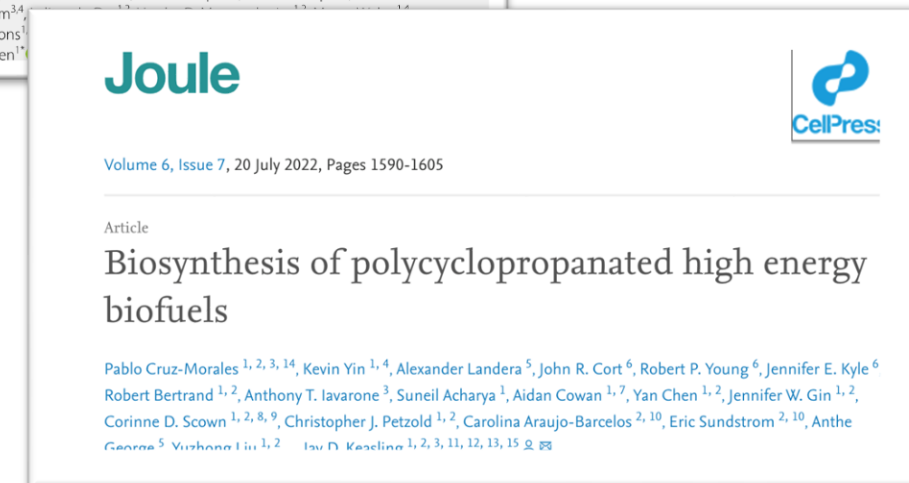
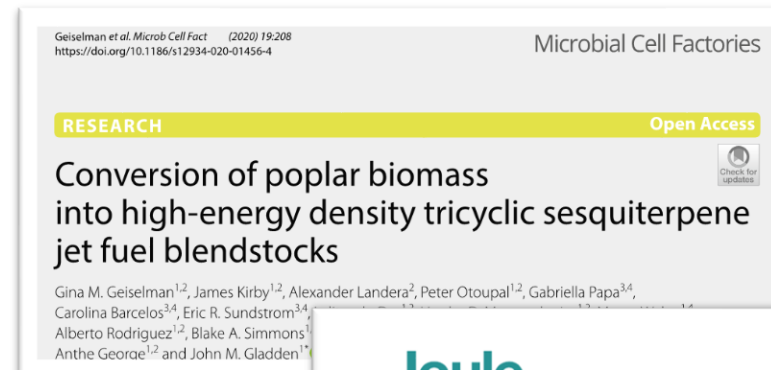
- Sesquiterpenes
- Polycyclopropanated molecules

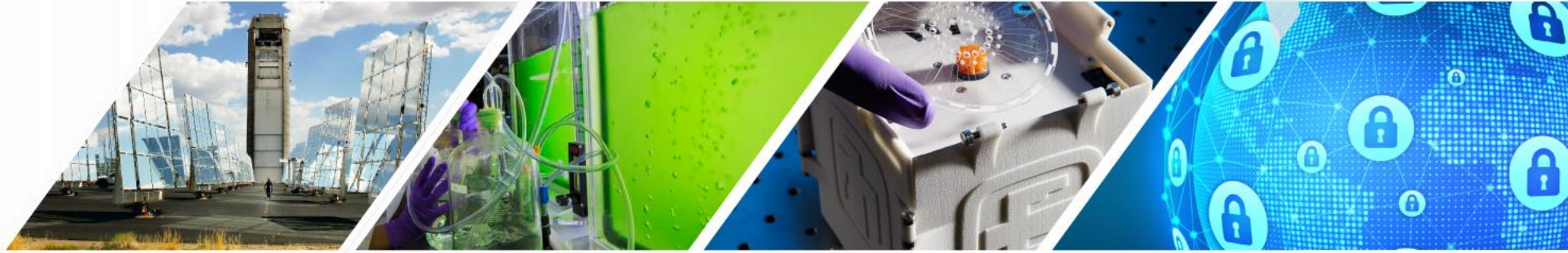
Blending models for fuel optimization:

- Accurate predictions of viscosity

Building Structure-Property Relationships of Cycloalkanes in Support of Their Use in Sustainable Aviation Fuels

Alexander Landera^{1*}, Ray P. Bambha¹, Najjia Hao², Sai Puneet Desai², Cameron M. Moore², Andrew D. Sutton^{2†} and Anthe George¹





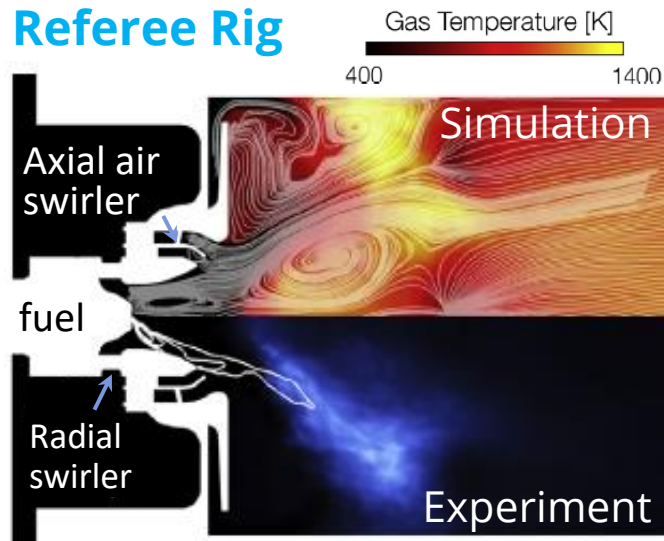
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End-Use

Isaac Ekoto
Manager, Applied Combustion Research



AFRL Referee Rig



Lean Blow-Out

code →	LESLIE	OpenNCC	FLUENT	VIDA	CONVERGE (All with PDDA spray inject & secondary breakup)					EXP
Fuel ↓	FR, PaSR, HyChem Reduced, PDDA w/ breakup	FR, Laminar, HyChem Reduced, PDDA no breakup	FR, EDM, 4-step tuned to HyChem, Rosin-R w/ breakup	FPV, PPDF, HyChem Detailed, Rosin-R w/ BU	Zonal FR, Laminar, Dryer-Won Compact	Zonal FR, Laminar, HyChem Skeletal	FR, Laminar, HyChem Detailed	Zonal FR, Laminar, HyChem Reduced	FGM Flamlet, PPDF, Dryer-Won Compact,	Average value, and +/- 2 st-dev
A-2	0.070	0.078	0.0885	.090	0.085	0.080	0.090	0.082	0.082	0.0806 0.0788 to 0.0824
C-1	0.074	0.087	0.094	.085	0.092	0.084	0.088	0.080	0.084	0.0869 0.08535 to 0.0884

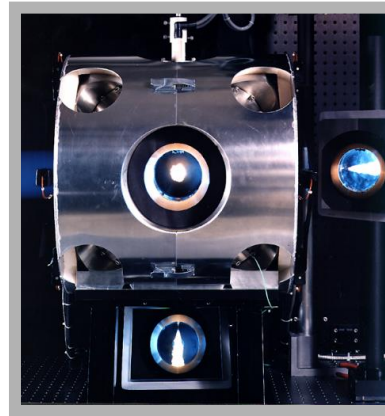
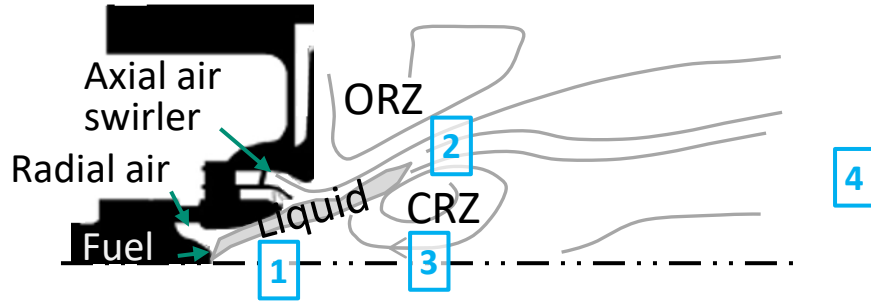
Book Chapter in "Fuel Effects on Operability of Aircraft Gas Turbine Combustors", Eds. M. Colket, J. Heyne, AIAA, 2021; ISBN 978-1-62410-603-3

- **Modeling Challenge:** Sub-models *lack modularity* and are *missing relevant processes*
- **Experimental Challenge:** Swirl stabilized combustors are *resource intensive* with *complex physicochemical interactions* and *uncertain boundary conditions*
- **Sandia Approach:** Interrogate relevant physics in *bespoke experiments* with *advanced optical and sampling diagnostics* used to obtain data needed for associated model development



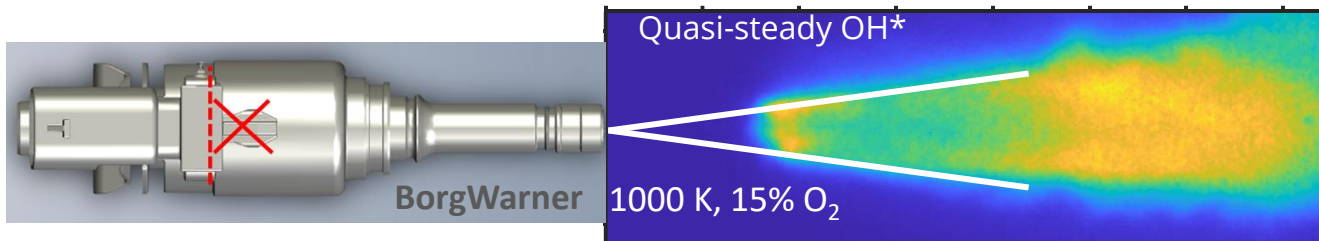
Rich, Quench, Lean (RQL) Combustor

1. **Liquid:** Spray breakup and atomization
2. **Flame:** Mixing and ignition
3. **Recirculation Zone:** Flame stabilization / soot formation
4. **Lean-Burn:** Soot aging and oxidation



Sandia Constant-Volume Spray Chamber

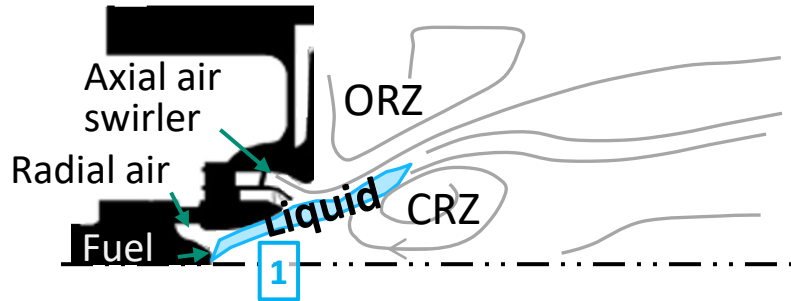
- Reproduces relevant CRZ conditions
 - 300 - 1800 K (by vitiation)
 - up to 350 bar
 - 0 - 21% O₂ (exhaust gas)



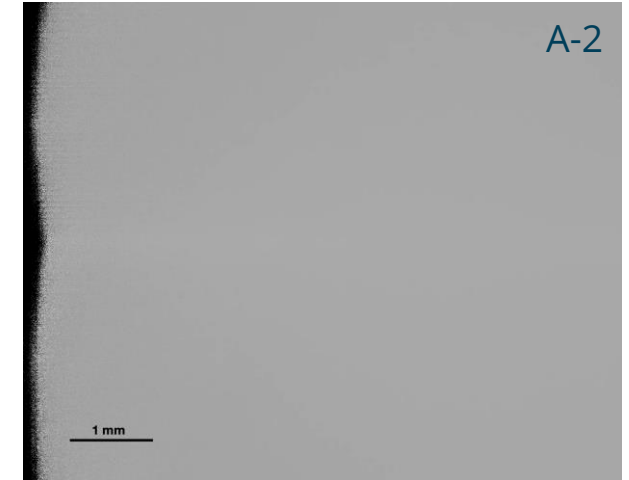
Single-Hole Atomizer

- Specialized injectors to produce relevant droplet sizes / velocities

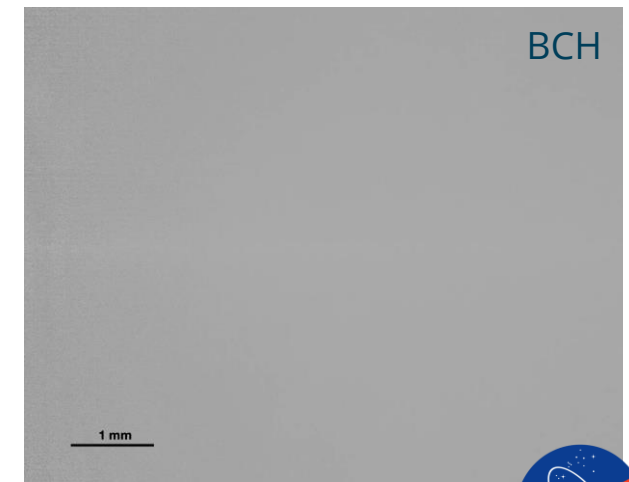
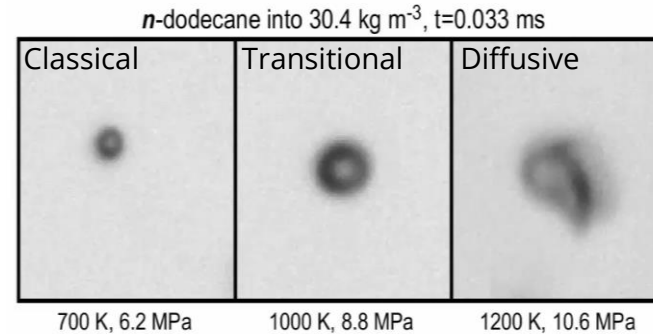
AT TAKEOFF CONDITIONS WHERE COMBUSTOR PRESSURES AND TEMPERATURES ARE ELEVATED, TRANSCRITICAL MIXING PROCESSES CAN DOMINATE



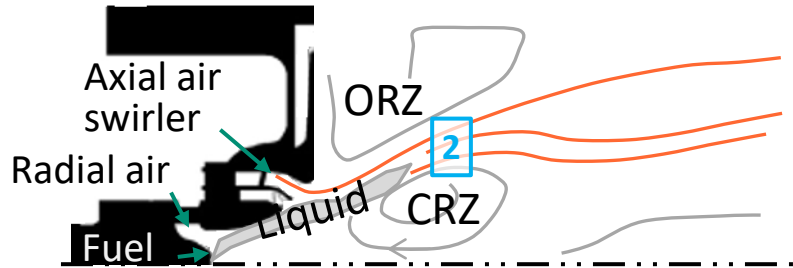
Fuel	Distillation
n-dodecane	Single component (216°C)
A-2 (Jet A)	Standard boiling
C-1	Narrow boiling
C-4	Wide boiling
Bicyclohexyl (BCH)	High boiling (227°C)



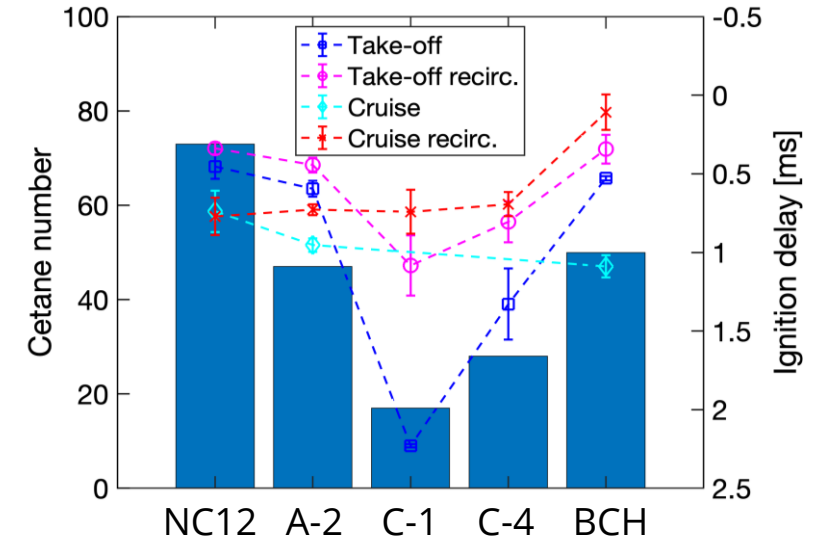
- Transcritical liquids undergo faster "diffusive" mixing due to the lack of a liquid-vapor interface



Current modeling approaches only consider classic droplet breakup and evaporation and thus will incorrectly predict the mixing state

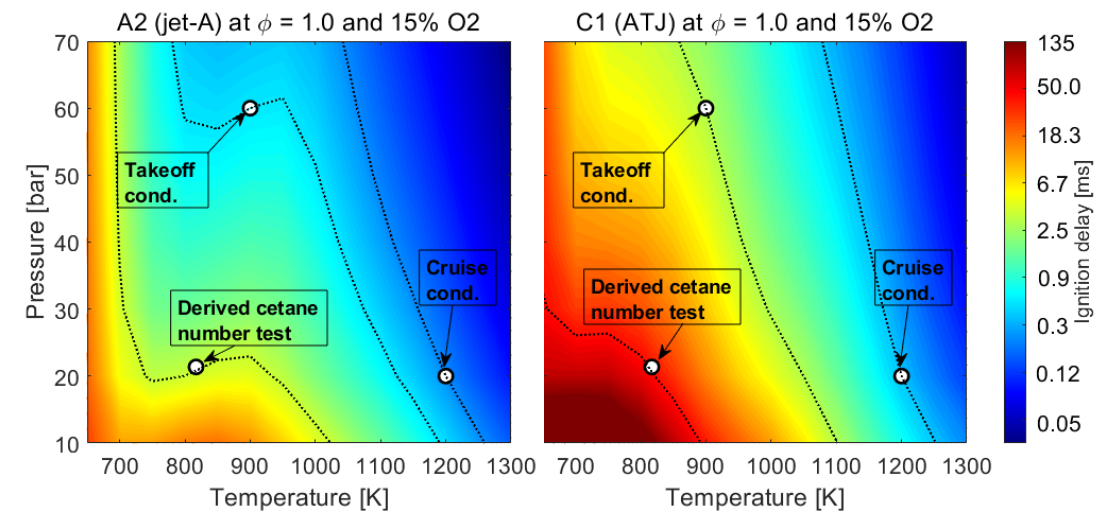


Combustor Condition	Temperature [K]	Pressure [bar]	O ₂ [%]
Cruise	800	20	21
Cruise (CRZ)	1200	20	15
Take-off	900	60	21
Take-off (CRZ)	1200	60	15
DCN	817	21.37	21

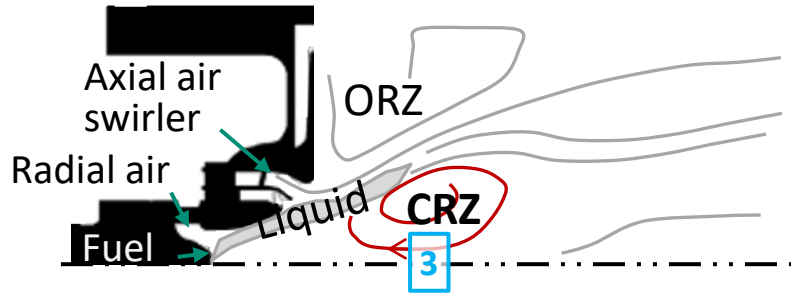


- **ASTM D4054:** Derived cetane number (DCN) is the only fuel combustion metric specified

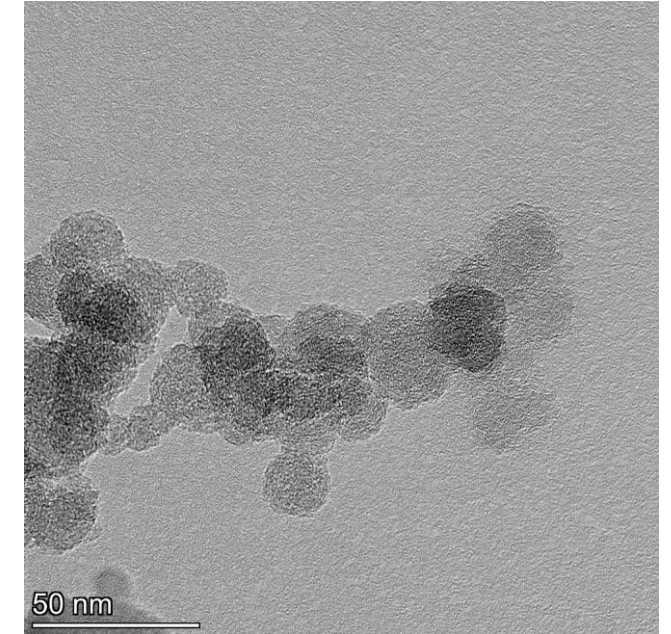
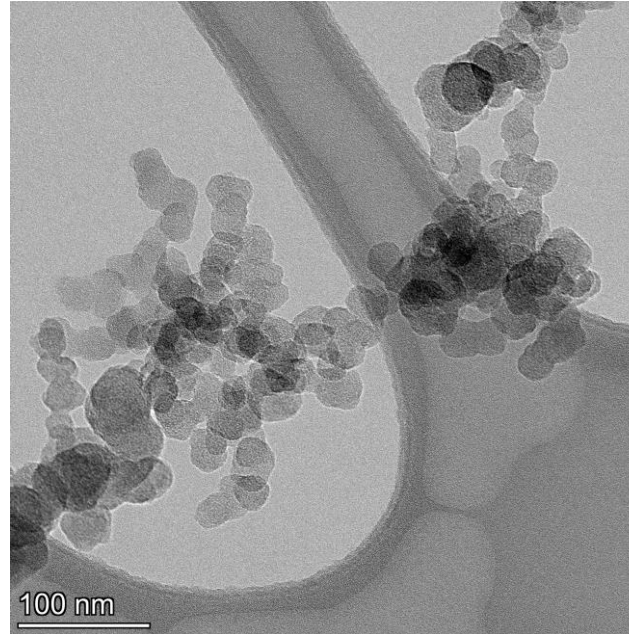
Ignition behavior at non-cruise conditions deviates substantially relative to correlations with DCN



Courtesy of Dario Lopez-Pintor



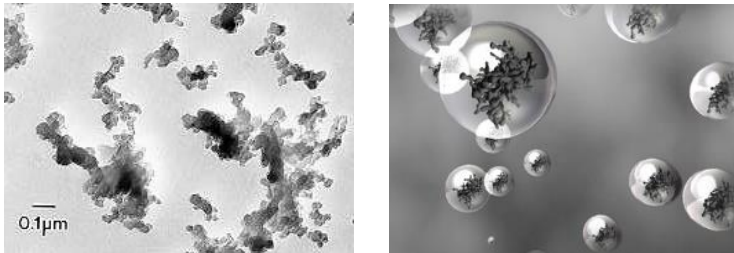
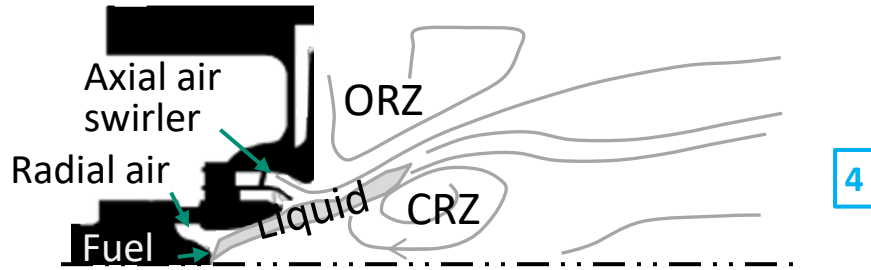
HR TEM



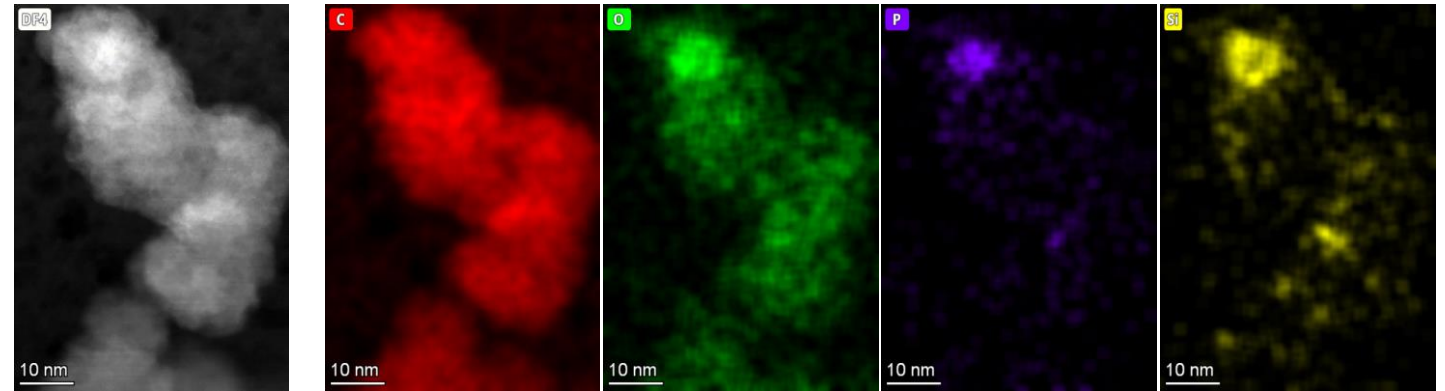
- **Electron Energy Loss Spectroscopy:** Soot particle density & bond structure

Soot sampling and high-resolution transmission electron microscopy (HR-TEM) used to characterize particle nanostructure and aggregate morphology

BEYOND SOOT PARTICLE MASS AND NUMBER, SURFACE PROPERTIES RELEVANT TO CONTRAILS FORMATION CAN ALSO BE CHARACTERIZED



- **Energy Dispersive X-ray Spectroscopy:** Soot surface atomic composition (e.g., sulfates)



- Companion project seeks to clarify water nucleation processes as a function of soot surface chemistry and morphology in a newly developed atmospheric chamber

Sandia approach complements existing gas turbine combustor research by leveraging well-controlled facilities that replicate relevant conditions to support development of physics-based modeling methods



Energy &
Homeland Security

Emissions & Repercussions

Shruti Mishra



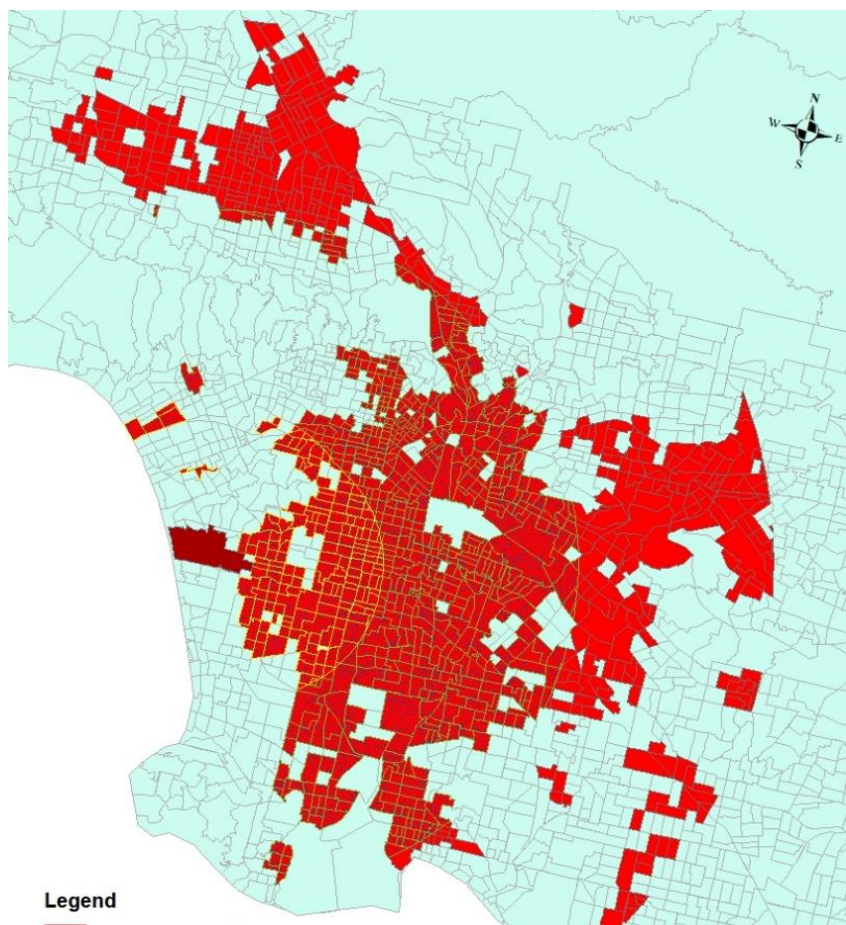
BENEFITS OF SUSTAINABLE AVIATION FUEL EXTEND BEYOND LOW CARBON

- The benefits of sustainable aviation fuel include co-benefits such as health benefits.
- Blending SAF into jet fuel could reduce the air pollutants due to lower sulfur and aromatics content in SAF (Benosa et al., 2018).
- Reduction in air pollutants lead to reduction in mortalities and morbidities related to the air pollutants (Arter et al., 2022).
- SNL are quantifying the health benefits of SAF including those to disadvantaged communities



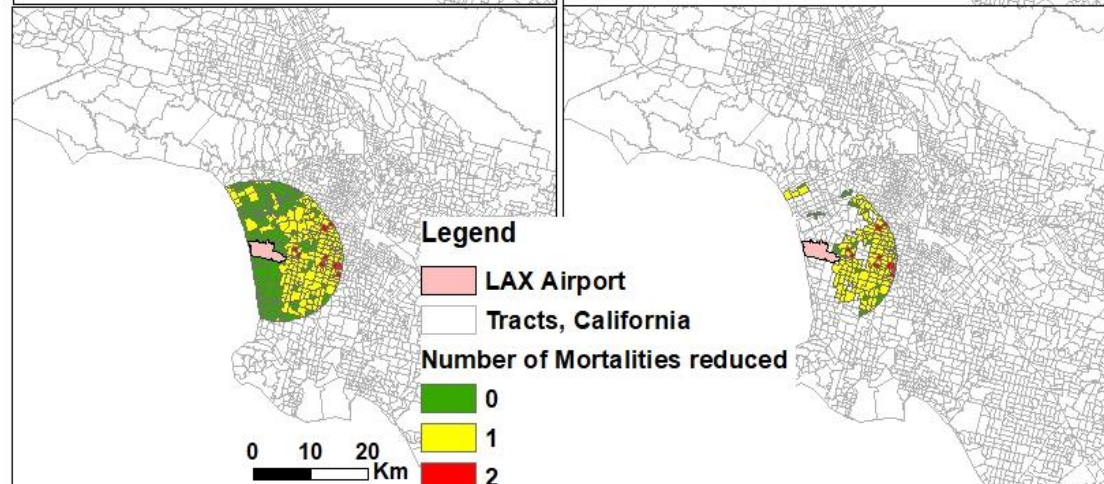
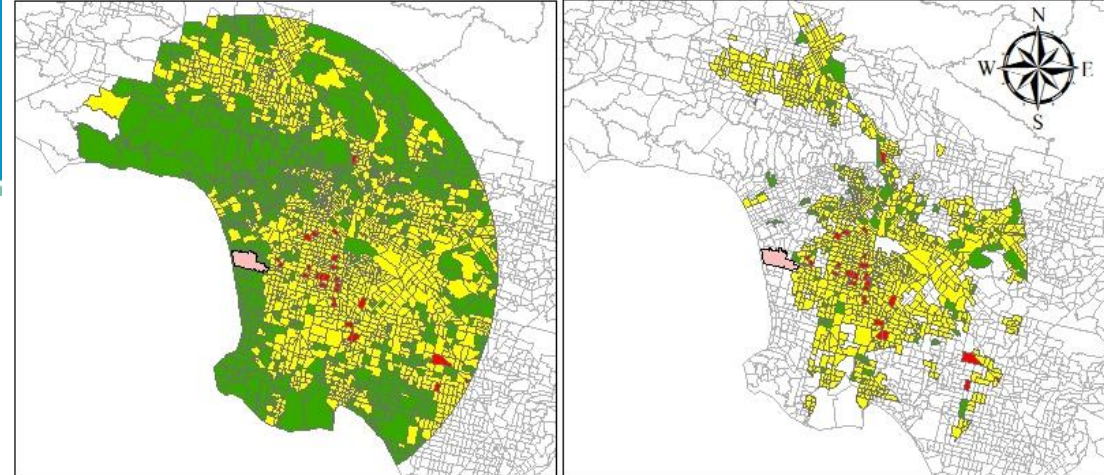


1. Air pollutants from Jet fuel Vs. Blended SAF/Jetfuel
2. Identify the area of influence (Aol)
3. Quantify the change in morbidity and mortalities due to increased use of SAF
4. Quantify the SAF led reduction in the number of mortalities among target disadvantaged communities (DAC).



Legend

- Los Angeles (LAX) Airport
- Disadvantaged Tracts within 10 KM of LAX
- Disadvantaged Tracts within 25 KM of LAX
- Disadvantaged Tracts within 40 KM of LAX
- Tracts surrounding Los Angeles (LAX) airport



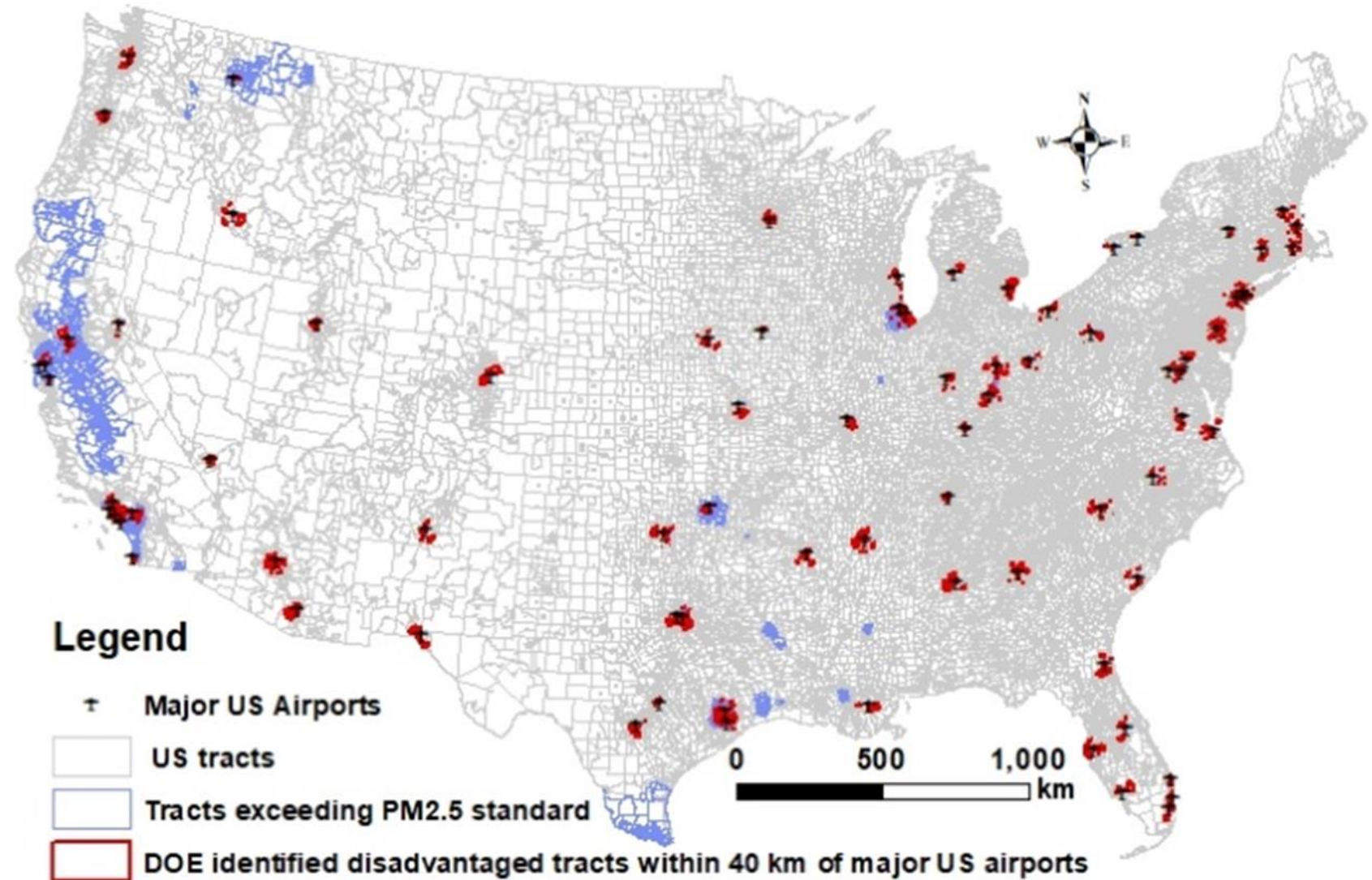
Legend

- LAX Airport
- Tracts, California
- Number of Mortalities reduced**
- 0
- 1
- 2

0 10 20
Km



The benefits of SAF to disadvantaged communities in other major airports in the U.S. should be significant.





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